

Lecture #2

OUTLINE

- Energy-band model
- Doping

Read: Chapter 2

Definition of Terms

n = number of electrons/cm³

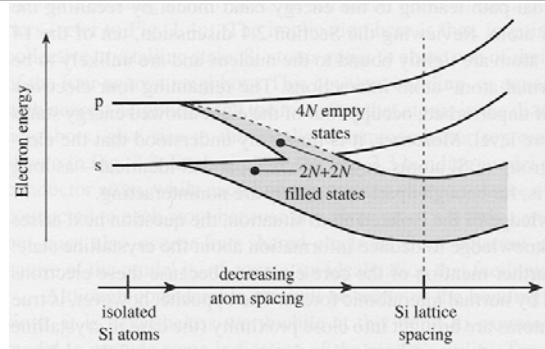
p = number of holes/cm³

n_i = intrinsic carrier concentration

In a pure semiconductor,

$$n = p = n_i$$

Si: From Atom to Crystal



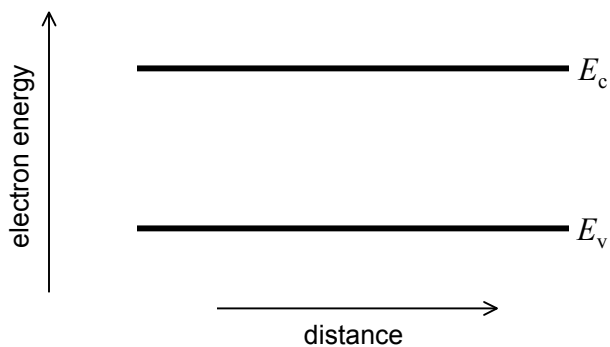
Energy states in Si atom → energy bands in Si crystal

- The highest nearly-filled band is the **valence band**
- The lowest nearly-empty band is the **conduction band**

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Energy Band Diagram



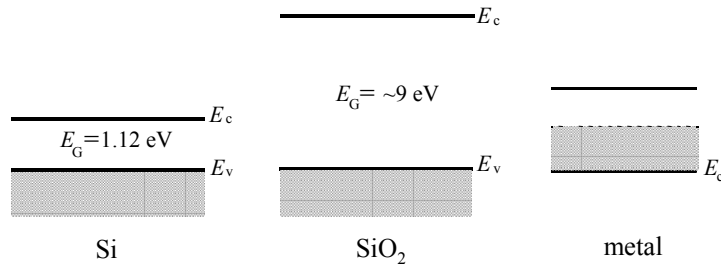
Simplified version of energy band model, indicating

- bottom edge of the conduction band (E_c)
- top edge of the valence band (E_v)
- E_c and E_v are separated by the **band gap energy E_g**

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Band Gap and Material Classification



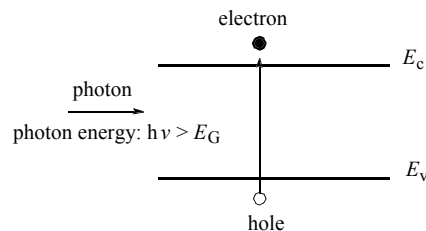
- Filled bands and empty bands do not allow current flow
- Insulators have large E_G
- Semiconductors have small E_G
- Metals have no band gap
 - conduction band is partially filled

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Measuring Band Gap Energy

- E_G can be determined from the minimum energy ($h\nu$) of photons that are absorbed by the semiconductor.



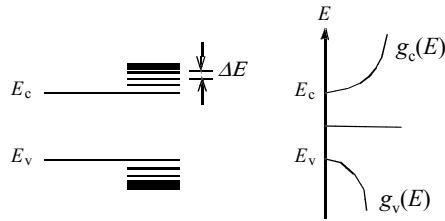
Band gap energies of selected semiconductors

Semiconductor	Ge	Si	GaAs
Band gap (eV)	0.67	1.12	1.42

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Density of States



$g(E)dE$ = number of states per cm^3 in the energy range between E and $E+dE$

Near the band edges:

$$g_c(E) = \frac{m_n^* \sqrt{2m_n^*(E - E_c)}}{\pi^2 \hbar^3} \quad E \geq E_c$$

$$g_v(E) = \frac{m_p^* \sqrt{2m_p^*(E_v - E)}}{\pi^2 \hbar^3} \quad E \leq E_v$$

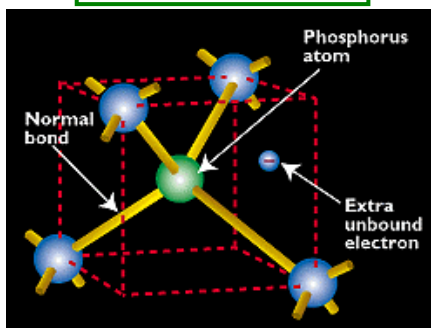
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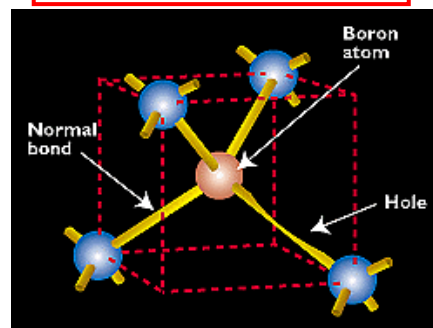
Doping

By **substituting** a Si atom with a special impurity atom (**Column V** or **Column III** element), a conduction electron or hole is created.

Donors: P, As, Sb



Acceptors: B, Al, Ga, In



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Doping Silicon with Donors

Example: Add arsenic (As) atom to the Si crystal

The loosely bound 5th valence electron of the As atom “breaks free” and becomes a mobile electron for current conduction.

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Doping Silicon with Acceptors

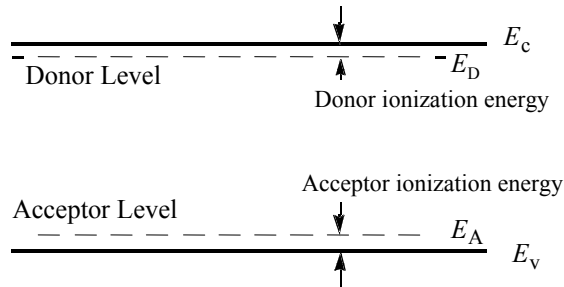
Example: Add boron (B) atom to the Si crystal

The B atom accepts an electron from a neighboring Si atom, resulting in a missing bonding electron, or “hole”. The hole is free to roam around the Si lattice, carrying current as a positive charge.

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Donor / Acceptor Levels (Band Model)



Ionization energy of selected donors and acceptors in silicon

Dopant	Donors			Acceptors		
	Sb	P	As	B	Al	In
Ionization energy, $E_c - E_d$ or $E_a - E_v$ (meV)	39	45	54	45	67	160

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Charge-Carrier Concentrations

N_D : ionized donor concentration (cm^{-3})

N_A : ionized acceptor concentration (cm^{-3})

Charge neutrality condition: $N_D + p = N_A + n$

At thermal equilibrium, $np = n_i^2$ ("Law of Mass Action")

$$n = \frac{N_D - N_A}{2} + \sqrt{\left(\frac{N_D - N_A}{2}\right)^2 + n_i^2}$$

Note: Carrier concentrations depend on *net* dopant concentration ($N_D - N_A$) !

$$p = \frac{N_A - N_D}{2} + \sqrt{\left(\frac{N_A - N_D}{2}\right)^2 + n_i^2}$$

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N-type Material

$$N_D \gg N_A$$

$$(N_D - N_A \gg n_i):$$

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P-type Material

$$N_A \gg N_D$$

$$(N_A - N_D \gg n_i):$$

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Terminology

donor: impurity atom that increases n

acceptor: impurity atom that increases p

n-type material: contains more electrons than holes

p-type material: contains more holes than electrons

majority carrier: the most abundant carrier

minority carrier: the least abundant carrier

intrinsic semiconductor: $n = p = n_i$

extrinsic semiconductor: doped semiconductor

Summary: Band Model

- Splitting of allowed atomic energy levels occurs in a crystal
 - Separation between energy levels is small, so we can consider them as bands of continuous energy levels
 - Highest nearly-filled band is the **valence band**
 - Lowest nearly-empty band is the **conduction band**

– Energy-band diagram:

- Shows only bottom edge of conduction band E_c and top edge of valence band E_v
- E_c and E_v are separated by the band-gap energy E_G
- Dopants introduce localized energy levels within the band gap:
 - donor level: slightly below E_c
 - acceptor level: slightly above E_v

Summary: Doping

- **Dopants in Si:**

- Reside on lattice sites (substituting for Si)
- Group V elements contribute conduction electrons, and are called **donors**
- Group III elements contribute holes, and are called **acceptors**
- Very low ionization energies (<50 meV)
 - ionized at room temperature

Dopant concentrations typically range from 10^{14} cm^{-3} to 10^{20} cm^{-3}